

**TETRACYCLIC COMPOUNDS FROM INDOLO[2,3-b] CYCLOHEPTAN-1-ONES.  
SYNTHESIS OF ISOXAZOLO[4,3':6,7]CYCLOHEPTA[b]INDOLES.**

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**Abstract**

Cycloheptan-1',2'-dione-1'-arylhydrazones **3** obtained from the Japp-Klingemann reaction of diazotised aniline derivatives **1** and 2-hydroxymethylene cycloheptanone **2** on acid catalysed cyclization afforded indolo[2,3-b]cycloheptan-1-ones **4**. These on mixed Aldol condensation with benzaldehyde followed by reaction with hydroxylamine hydrochloride gave isoxazolo[4,3':6,7]cyclohepta[b]indoles **6**.

**Introduction**

Indole derivatives were reported to have pharmacological activities such as antitumour<sup>1</sup>, antiinflammatory<sup>2</sup>, antibacterial<sup>3</sup>, antifungal<sup>4</sup>, antituberculosis<sup>4</sup> activities. Besides this, the simple cyclooctane ring fused with indole [Iprindole] was found to have antidepressant property<sup>4</sup>. In the above context our present investigation was aimed at to synthesise the hitherto unknown cycloheptane ring fused with indoles, indolo[2,3-b]cycloheptan-1-ones<sup>5</sup> from the diazotised anilines **1** and 2-hydroxymethylenecycloheptanone **2**<sup>6</sup> which were used as synthons to derive isoxazolo[4,3':6,7] cyclohepta[b]indoles **6** (scheme 1).

**Experimental**

**General information**

Thin layer chromatography was used to access reactions and purity of products. Preparative separation was performed in columns packed with silica gel. Melting points were determined in a Mettler Fb - 5 apparatus and are uncorrected. The IR spectra were recorded in KBr pellets with a shimadzu FTIR - 8201 (PC)s spectrometer and only noteworthy absorption levels (reciprocal centimeter) are listed. The <sup>1</sup>H NMR spectra were recorded on varian AMX 400 spectrometer. Chemical shifts are reported in parts per million (δ) downfield from the internal standard, tetramethylsilane (Me<sub>4</sub> Si). Signal multiplicities are represented by s (singlet), d (doublet), b s (broad singlet) and m (multiplet). Satisfactory microanalyses were obtained on Carlo Erba 1106 and Perkin Elmer model 240 CHN analyzers.

**Preparation of 2-hydroxymethylene cycloheptanone 2**

Cycloheptanone (2.35 mL, 0.02 mol) was added in portions over a period of five minutes to a well cooled, vigorously stirred mixture of sodium methoxide (from 5.17 g of sodium in 5 mL of absolute methanol), dry ether (4 mL) and ethyl formate (1.8 mL, 0.02 mol). The mixture was stirred in the ice bath for another one hour and then allowed to stand at room temperature for 24 hours. At the end of the period, ice and water were added to the yellow solid mass and acidified with concentrated hydrochloric acid. The oil that separated was extracted with ether, washed with cold water and brine and dried over anhydrous sodium sulphate. The residual oil after the removal of solvent was distilled under reduced pressure to give 2-hydroxymethylene cycloheptanone 2 as viscous liquid with 70% yield.

**General procedures****Preparation of cycloheptan-1',2'-dione-1'-arylhydrazones 3**

A mixture of 2-hydroxymethylenecycloheptanone 2 (0.004 mol) and sodium acetate trihydrate (1g) in methanol (6 mL) was cooled in ice. A solution of appropriate aniline derivative (0.004 mol) in aqueous hydrochloric acid (1.8 mL of HCl in 2.12 mL H<sub>2</sub>O) was diazotised with cold saturated solution of sodium nitrite (0.35 g in 0.8 mL water) between 0°C and -5°C. The diazotised solution was added in small portions to the ice cooled mixture containing 2-hydroxymethylenecycloheptanone 2 over a period of 0.5 hours with constant stirring. After standing for 0.5 hours more, the resulting solid was filtered, washed with water dried and crystallised from ethanol. The physical and spectral data of the hydrazones 3 were presented in Table 1.

**Cyclization of the hydrazones to indolo[2,3-*b*]cycloheptan-1-ones 4**

The appropriate hydrazone 3 (0.001 mol) in a mixture of acetic acid (20 mL) and concentrated hydrochloric acid (5 mL) was refluxed on oil bath pre-heated to 125-130°C for 2 hours. The contents were then cooled and poured into ice water with stirring. The separated brown solid was filtered and purified by passing through a column of silica gel and eluting with petroleum ether - ethyl acetate (95:5) mixture. The physical and spectral data of all the compounds 4a-e were given in Table 2.

**Mixed Aldol condensation of indolo[2,3-*b*]cycloheptan-1-ones 4 with benzaldehyde.**

A mixture of Indolo[2,3-*b*]cycloheptan-1-ones 4 (0.001 mol) and benzaldehyde (0.001 mol) was treated with 4% alcoholic potassium hydroxide (10 mL) and stirred for 12 hours at room temperature. The precipitated crystalline product was filtered off and washed with 50% ethanol. A further crop of condensation product was obtained on neutralisation with acetic acid and dilution with water. The products 5 were crystallised from methanol and their physical and spectral data were presented in Table 3.

**Synthesis of 3-phenyl isoxazolo[4',3':6,7] cyclohepta[*b*]indoles 6**

A mixture of 2-benzylideneindolo[2,3-*b*]cycloheptan-1-one 5 (0.001 mol), hydroxylamine hydrochloride (1.5 g) in dry pyridine (5 mL) was heated at 130°C for 10 hours. The reaction mixture was then poured into crushed ice, the resulting semi solid separated was extracted with chloroform and washed successively with dilute hydrochloric acid and water, dried over anhydrous sodium sulphate. Evaporation of the solvent yielded a crude product which was purified by

passing through a silica gel column and eluting with petroleum ether-ethyl acetate (98:2) mixture to afford the product **6**. The physical and spectral data were given in Table 4.

### Results and Discussion

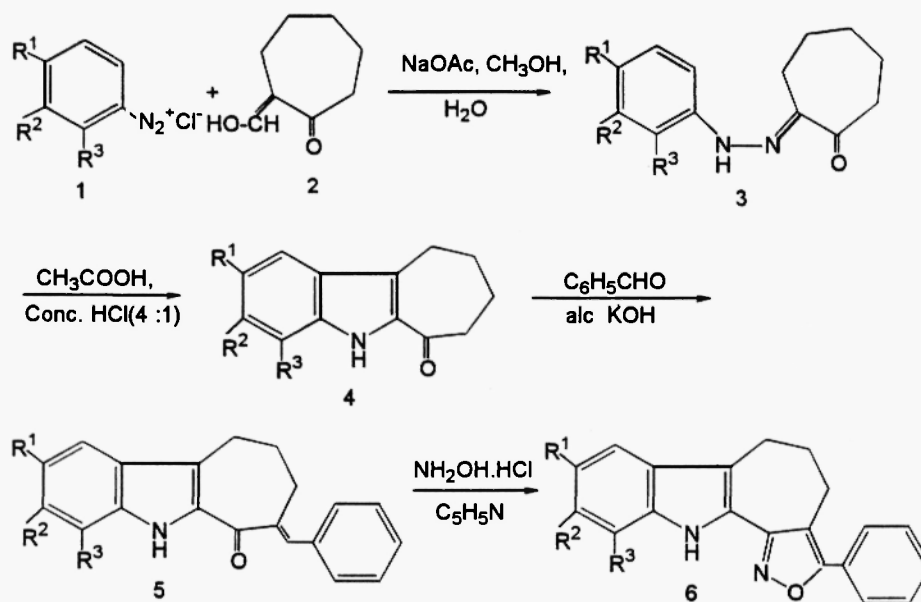
The Japp-Klingemann reaction<sup>9</sup> of diazonium salt solution of p-toluidine derivative **1a** with 2-hydroxymethylenecycloheptanone **2** gave the product **3a** m.p. 85°C which showed the bands at 1654 cm<sup>-1</sup> due to carbonyl stretching and 1624 cm<sup>-1</sup> due to C=N stretching in the IR spectrum. Its proton NMR spectrum registered a singlet at  $\delta$  2.30 due to a methyl group at C<sub>4</sub>. The C<sub>3</sub>, C<sub>4</sub>, C<sub>5</sub>, C<sub>6</sub> and C<sub>7</sub> protons resonate as a multiplet at  $\delta$  1.77-2.68. The resonance signal corresponding to four aromatic protons appeared as a multiplet at  $\delta$  7.05-7.26 and a singlet at  $\delta$  13.67 for NH proton. The above spectrum clearly indicates that the structure of the compound is **3a**. The elemental analysis of the compound agreed very well with the proposed molecular formula C<sub>14</sub>H<sub>18</sub>N<sub>2</sub>O augmenting the structure of the compound to be **3a**.

The appropriate hydrazone **3a** upon acid cyclization using Kent's reagent (acetic acid / HCl, 4:1) gave the product **4a** which showed the bands at 1618 cm<sup>-1</sup> due to carbonyl group and at 3328 cm<sup>-1</sup> for NH stretching and its <sup>1</sup>H NMR spectrum displayed a singlet at  $\delta$  2.45 for C<sub>7</sub> - CH<sub>3</sub>, three aromatic protons envelop at  $\delta$  7.16-7.42 (C<sub>6</sub>, C<sub>8</sub>, C<sub>9</sub> protons). C<sub>5</sub> and C<sub>2</sub> protons appeared as two multiplets at  $\delta$  3.11-3.14 and  $\delta$  2.82 - 2.85, respectively. C<sub>4</sub> and C<sub>3</sub> protons resonate as multiplets at  $\delta$  2.05-2.11 and  $\delta$  1.95 - 2.02, respectively. The appearance of NH proton as a broad singlet at  $\delta$  8.86 clearly indicates that the product **3a** cyclised to **4a**. The elemental analysis of the compound **4a** agreed very well with the proposed molecular formula C<sub>14</sub>H<sub>15</sub>NO, thereby confirming the structure **4a** for the compound formed. A series of similar indolo[2,3-b]cycloheptan-1-ones **4b-e** were realised from **1b-e** and **2** through the corresponding hydrazones **3b-e**.

Mixed Aldol condensation<sup>7</sup> of indolo[2,3-b]cycloheptan-1-one **4a** with benzaldehyde under basic condition leads to the formation of 2-benzylideneindolo[2,3-b]cycloheptan-1-one **5a**. The structure of **5a** was established on the basis of elemental and spectral data. The IR spectrum exhibited a strong absorption band at 1641 cm<sup>-1</sup> characteristic of  $\alpha$ ,  $\beta$  - unsaturated carbonyl group and a band at 3311 cm<sup>-1</sup> ascribable to NH group. In its <sup>1</sup>H NMR spectra the disappearance of C<sub>2</sub> protons signal and appearance of benzylic proton signal as a singlet at  $\delta$  7.78 proved the mixed Aldol condensation of **4a** with benzaldehyde to give **5a**. The aromatic cluster accounting for eight protons appeared at  $\delta$  7.18-7.43 as a multiplet. The C<sub>3</sub> and C<sub>5</sub> protons appeared as two multiplets at  $\delta$  2.89-2.91 and  $\delta$  3.18 -3.21, respectively. The C<sub>4</sub> protons resonate as a multiplet at  $\delta$  2.17-2.23 while that of methyl group and indole NH as a singlet at  $\delta$  2.46 and as a broad singlet at  $\delta$  8.99, respectively. The elemental analysis : C 83.70, H 06.39 and N 04.59 is compatible with the molecular formula C<sub>21</sub>H<sub>19</sub>NO. A series of similar compounds was realised with **4b**, **4c**, **4d** and **4e**.

The reaction of 2-benzylideneindolo[2,3-b]cycloheptan-1-one **5a-e** with hydroxylamine hydrochloride in pyridine gave the desired 3-phenyl isoxazolo[4',3':6,7] cyclohepta[b]indoles<sup>8</sup> **6a-e** in moderate yields. The IR spectrum of **6a** revealed absorption band around 1600 cm<sup>-1</sup> indicating the transformation of **5a** to **6a**. The <sup>1</sup>H NMR spectrum of **6a** in CDCl<sub>3</sub> showed a multiplet at  $\delta$  2.12 - 2.17 for C<sub>5</sub> protons. The eight aromatic protons appeared as a complex multiplet at  $\delta$  7.09-7.73. The methyl group protons and indole NH proton exhibited a singlet at  $\delta$  2.47 and a broad singlet at  $\delta$  8.74, respectively. The elemental analysis agreed well with the molecular formula C<sub>21</sub>H<sub>18</sub>N<sub>2</sub>O.

Scheme 1



- 1, 2-6 a :  $\text{R}^1 = \text{CH}_3, \text{R}^2 = \text{R}^3 = \text{H}$   
 b :  $\text{R}^1 = \text{R}^3 = \text{H}, \text{R}^2 = \text{CH}_3$   
 c :  $\text{R}^1 = \text{R}^2 = \text{H}, \text{R}^3 = \text{CH}_3$   
 d :  $\text{R}^1 = \text{R}^2 = \text{R}^3 = \text{H}$   
 e :  $\text{R}^1 = \text{Cl}, \text{R}^2 = \text{R}^3 = \text{H}$

Table 1. Physical and spectral data of cycloheptan-1'2'-dione-1'-aryl hydrazones **3**

Compound	M.P. (°C) Solvent	Yield (%)	IR (ν)	Molecular Formula	Analysis (%)	
					Calcd	Found
<b>3a</b>	85 PE-EA	79	3271 1654, 1624	C <sub>13</sub> H <sub>18</sub> N <sub>2</sub> O (230.31)	C	73.01 (73.00)
					H	07.88 (07.98)
					N	12.16 (12.10)
<b>3b</b>	73 PE-EA	72	3286 1633, 1604	C <sub>14</sub> H <sub>18</sub> N <sub>2</sub> O (230.31)	C	73.01 (73.07)
					H	07.88 (07.90)
					N	12.16 (12.20)
<b>3c</b>	90 PE-EA	69	3282 1629, 1602	C <sub>14</sub> H <sub>18</sub> N <sub>2</sub> O (230.31)	C	73.01 (73.13)
					H	07.88 (07.98)
					N	12.16 (12.18)
<b>3d</b>	87 PE-EA	75	3290 1596, 1601	C <sub>13</sub> H <sub>16</sub> N <sub>2</sub> O (216.28)	C	72.19 (72.28)
					H	07.46 (07.40)
					N	12.95 (12.91)
<b>3e</b>	105 PE-EA	77	3289 1627, 1595	C <sub>13</sub> H <sub>17</sub> N <sub>2</sub> OCl (250.73)	C	62.28 (62.30)
					H	06.03 (06.10)
					N	11.17 (11.19)

PE: Petroleum ether : 60 - 80°C, EA : Ethyl acetate

Table 3. Physical and spectral data of 2-benzylidene indolo[2,3-b]cycloheptan-1-ones **5**

Compound	M.P. (°C) Solvent	Yield (%)	IR (ν)	Molecular Formula	Analysis (%)	
					Calcd	Found
<b>5a</b>	182.4 PE-EA	80	3311 1641, 1568	C <sub>21</sub> H <sub>19</sub> NO (301.39)	C	83.69 (83.70)
					H	06.35 (06.39)
					N	04.65 (04.59)
<b>5b</b>	184.9 PE-EA	87	3315 1641, 1568	C <sub>21</sub> H <sub>9</sub> NO (301.39)	C	83.69 (83.78)
					H	06.35 (06.28)
					N	04.65 (04.73)
<b>5c</b>	165.3 PE-EA	83	3302 1637, 1577	C <sub>21</sub> H <sub>19</sub> NO (301.39)	C	83.69 (83.58)
					H	06.35 (06.25)
					N	04.65 (04.76)
<b>5d</b>	161 PE-EA	79	3311 1635, 1566	C <sub>20</sub> H <sub>17</sub> NO (287.36)	C	83.59 (83.55)
					H	05.96 (05.98)
					N	04.88 (04.80)
<b>5e</b>	181 PE-EA	90	3340 1636, 1579	C <sub>20</sub> H <sub>16</sub> NOCl (321.81)	C	74.65 (74.69)
					H	05.01 (05.15)
					N	04.35 (04.30)

PE: Petroleum ether : 60 - 80°C, EA : Ethyl acetate

Table 2. Physical and spectral data of indol[2,3-*b*]cycloheptan-1-ones 4

Compound	M.P. (°C) Solvent	Yield (%)	IR (ν)	Molecular Formula	Analysis (%)		<sup>1</sup> H - NMR
					Calcd	Found	
4a	178 P3-EA	64	3328 1618	C <sub>14</sub> H <sub>13</sub> NO (213.28)	C 78.84	(78.82)	1.95-2.92 (m, 2H, C <sub>1</sub> -H <sub>2</sub> ), 2.05-2.11 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 2.45 (s, 3H, C <sub>5</sub> -CH <sub>3</sub> ), 2.82 - 2.85 (m, 2H, C <sub>2</sub> -H <sub>2</sub> ), 3.11-3.14 (m, 2H, C <sub>3</sub> -H <sub>2</sub> ), 7.16-7.42 (m, 3H, C <sub>6</sub> -H, C <sub>8</sub> -H, C <sub>9</sub> -H) and 8.86 (b s, 1H, NH)
					H 07.09	(07.15)	
					N 06.57	(06.45)	
4b	120 PE-EA	67	3329 1610	C <sub>14</sub> H <sub>13</sub> NO (213.28)	C 78.84	(78.91)	1.91-2.01 (m, 2H, C <sub>3</sub> -H <sub>2</sub> ), 2.06-2.11 (m, 2H, C <sub>1</sub> -H <sub>2</sub> ), 2.46 (s, 3H, C <sub>5</sub> -CH <sub>3</sub> ), 2.77-2.84 (m, 2H, C <sub>2</sub> -H <sub>2</sub> ), 3.11-3.14 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 6.81-7.54 (m, 3H, C <sub>6</sub> -H, C <sub>7</sub> -H, C <sub>8</sub> -H) and 8.77 (b s, 1H, NH)
					H 07.09	(07.23)	
					N 06.57	(06.82)	
4c	139-140 PE-EA	65	3298 1605	C <sub>14</sub> H <sub>13</sub> NO (213.28)	C 78.84	(78.94)	1.97-2.02 (m, 2H, C <sub>3</sub> -H <sub>2</sub> ), 2.06-2.13 (m, 2H, C <sub>1</sub> -H <sub>2</sub> ), 2.49 (s, 3H, C <sub>5</sub> -CH <sub>3</sub> ), 2.82-2.86 (m, 2H, C <sub>2</sub> -H <sub>2</sub> ), 3.13-3.17 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 7.03 - 7.67 (m, 3H, C <sub>6</sub> -H, C <sub>7</sub> -H, C <sub>8</sub> -H) and 8.84 (b s, 1H, NH)
					H 07.09	(07.21)	
					N 06.57	(06.37)	
4d	151 P3-EA	61	3292 1635	C <sub>13</sub> H <sub>11</sub> NO (199.25)	C 78.36	(78.43)	1.99-2.04 (m, 2H, C <sub>1</sub> -H <sub>2</sub> ), 2.06-2.12 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 2.82-2.87 (m, 2H, C <sub>2</sub> -H <sub>2</sub> ), 3.14 - 3.17 (m, 2H, C <sub>3</sub> -H <sub>2</sub> ), 7.11-7.67 (m, 4H, C <sub>6</sub> -H, C <sub>7</sub> -H, C <sub>8</sub> -H, C <sub>9</sub> -H) and 8.96 (b s, 1H, NH)
					H 06.58	(06.49)	
					N 07.03	(07.10)	
4e	172 PE-EA	60	3325 1610	C <sub>13</sub> H <sub>11</sub> NO (233.69)	C 66.81	(66.78)	1.96-2.06 (m, 2H, C <sub>1</sub> -H <sub>2</sub> ), 2.08-2.12 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 2.83-2.86 (m, 2H, C <sub>2</sub> -H <sub>2</sub> ), 3.08-3.11 (s, 2H, C <sub>3</sub> -H <sub>2</sub> ), 7.26-7.31 (m, 2H, C <sub>6</sub> -H, C <sub>7</sub> -H, C <sub>8</sub> -H), and 8.99 (b s, 1H, NH)
					H 06.00	(06.03)	
					N 05.17	(05.27)	

PE: Petroleum ether : 60 - 80°C, EA: Ethyl acetate

Table 4. Physical and spectral data of 3-phenyl isoxazolol[4',3':6,7]cyclohepta[b]indoles 6

Compound	M.P. (°C) Solvent	Yield (%)	IR ( $\nu$ )	Molecular Formula	Analysis (%) Calc Found	<sup>1</sup> H-NMR
6a	227 PE-EA	45	3284 1600	C <sub>21</sub> H <sub>18</sub> N <sub>2</sub> O (314.39)	C 80.23 (80.32) H 05.77 (05.65) N 08.91 (08.88)	2.12-2.17 (m, 2H, C <sub>5</sub> -H <sub>2</sub> ), 2.47 (s, 3H, C <sub>8</sub> -CH <sub>3</sub> ), 3.06-3.08 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 3.16-3.19 (m, 2H, C <sub>6</sub> -H <sub>2</sub> ), 7.09-7.73 (m, 8H, C <sub>7</sub> -H, C <sub>9</sub> -H, C <sub>10</sub> -H, C <sub>2</sub> -H, C <sub>1</sub> -H, C <sub>3</sub> -H, C <sub>5</sub> -H, C <sub>6</sub> -H) and 8.74 (b s 1H, NH)
6b	197.2 PE-EA	42	3250 1558	C <sub>21</sub> H <sub>18</sub> N <sub>2</sub> O (314.39)	C 80.23 (80.53) H 05.77 (05.98) N 08.91 (08.88)	2.13-2.27 (m, 2H, C <sub>5</sub> -H <sub>2</sub> ), 2.48 (s, 3H, C <sub>8</sub> -CH <sub>3</sub> ), 3.05-3.08 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 3.16-3.19 (m, 2H, C <sub>6</sub> -H <sub>2</sub> ), 6.83-7.73 (m, 8H, C <sub>7</sub> -H, C <sub>9</sub> -H, C <sub>10</sub> -H, C <sub>2</sub> -H, C <sub>1</sub> -H, C <sub>3</sub> -H, C <sub>5</sub> -H, C <sub>6</sub> -H) and 8.69 (b s, 1H, NH)
6c	210 PE-EA	55	3200 1558	C <sub>21</sub> H <sub>18</sub> N <sub>2</sub> O (314.39)	C 80.23 (80.22) H 05.77 (05.47) N 08.91 (08.82)	2.14-2.17 (m, 2H, C <sub>5</sub> -H <sub>2</sub> ), 2.55 (s, 3H, C <sub>8</sub> -CH <sub>3</sub> ), 2.94-2.97 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 3.18-3.20 (m, 2H, C <sub>6</sub> -H <sub>2</sub> ), 7.07-7.69 (m, 8H, C <sub>7</sub> -H, C <sub>9</sub> -H, C <sub>10</sub> -H, C <sub>2</sub> -H, C <sub>1</sub> -H, C <sub>3</sub> -H, C <sub>5</sub> -H, C <sub>6</sub> -H) and 8.57 (b s, 1H, NH)
6d	207.6 PE-EA	49	3274 1598	C <sub>20</sub> H <sub>15</sub> N <sub>2</sub> O (300.36)	C 79.98 (79.62) H 05.37 (05.42) N 09.33 (09.21)	2.13-2.18 (m, 2H, C <sub>5</sub> -H <sub>2</sub> ), 3.07-3.10 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 3.19-3.22 (m, 2H, C <sub>6</sub> -H <sub>2</sub> ), 7.12-7.73 (m, 9H, C <sub>7</sub> -H, C <sub>9</sub> -H, C <sub>10</sub> -H, C <sub>2</sub> -H, C <sub>1</sub> -H, C <sub>3</sub> -H, C <sub>5</sub> -H, C <sub>6</sub> -H) and 8.82 (b s, 1H, NH)
6e	235 PE-EA	40	3277 1508	C <sub>20</sub> H <sub>15</sub> N <sub>2</sub> OCl (334.80)	C 71.75 (71.65) H 04.51 (04.55) N 08.37 (08.36)	2.11-2.17 (m, 2H, C <sub>5</sub> -H <sub>2</sub> ), 3.06-3.09 (m, 2H, C <sub>4</sub> -H <sub>2</sub> ), 3.13-3.16 (m, 2H, C <sub>6</sub> -H <sub>2</sub> ), 7.20-7.73 (m, 8H, C <sub>7</sub> -H, C <sub>9</sub> -H, C <sub>10</sub> -H, C <sub>2</sub> -H, C <sub>1</sub> -H, C <sub>3</sub> -H, C <sub>5</sub> -H, C <sub>6</sub> -H) and 8.92 (b s, 1H, NH)

PE: Petroleum ether : 60 - 80°C, EA : Ethyl acetate

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